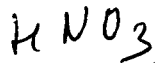
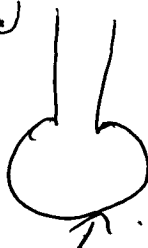


27



10 mL

64%, 1.45 g/mL



$$V_D = 100 \text{ mL}$$

$$M = \frac{9.78 / 63}{0.1} = 1.47 \text{ M}$$

Soluto $\Rightarrow m_D = 14.5 \text{ g}$
 $m_S = 64\% \text{ de } 14.5 = 9.28 \text{ g Soluto.}$